

The Cephalostatins. 22. Synthesis of Bis-steroidal Pyrazine Pyrones¹

George R. Pettit,* Bryan R. Moser,† Ricardo F. Mendonça, John C. Knight, and Fiona Hogan

Cancer Research Institute and Department of Chemistry and Biochemistry, Arizona State University, P.O. Box 871604, Tempe, Arizona 85287-1604, United States

Supporting Information

ABSTRACT: Cephalostatin 1 (1), a remarkably strong cancer cell growth inhibitory trisdecacyclic, bis-steroidal pyrazine isolated from the marine tube worm *Cephalodiscus gilchristi*, continues to be an important target for practical total syntheses and a model for the discovery of less complex structural modifications with promising antineoplastic activity. In the present study, the cephalostatin E and F rings were greatly simplified by replacement at C-17 with an *α*-pyrone (in 12), typical of the steroidal bufodienolides, and by a dihydro-*γ*-pyrone (in 16). The synthesis of pyrazine 12 from 5*α*-

dihydrotestosterone (nine steps, 8% overall yield) provided the first route to a bis-bufadienolide pyrazine. Dihydro- γ -pyrone 16 was synthesized in eight steps from ketone 13. While only insignificant cancer cell growth inhibitory activity was found for pyrones 12 and 16, the results provided further support for the necessity of more closely approximating the natural D–F ring system of cephalostatin 1 in order to obtain potent antineoplastic activity.

The isolation and structure of cephalostatin 1 (1), a powerful anticancer constituent of the Indian Ocean colonial marine tube worm Cephalodiscus gilchristi Ridewood (Cephalodiscidae), was summarized by us^{2a} 24 years ago. Subsequently, 1 became the prototype of the cephalostatins² and ritterazines, which together make up a unique family of 45 highly oxygenated bis-steroidal pyrazines.3 These marine invertebrate constituents exhibit powerful cancer cell growth inhibitory behavior (e.g., murine P388 lymphocytic leukemia cell line, ED₅₀ $10^{-7} \mu g/mL$; NCI 60 human cancer cell lines, GI₅₀ 1.8 nM).² The availability of the cephalostatins and ritterazines from their only known natural sources, C. gilchristi and the marine tunicate Ritterella tokioka, is still extremely limited. As a result, in vivo anticancer evaluation of these very promising natural products and subsequent preclinical development have been greatly restricted.

The outstanding antineoplastic potency together with the new and challenging molecular architecture of 1 and poor availability from natural sources ($\sim 10^{-6}$ % yields from C. gilchristi) soon led to synthetic approaches by a number of research groups. By 1998, Fuchs and colleagues had completed the first total synthesis of 1 (in 65 synthetic steps)^{4a} and of two additional members of the cephalostatin family (7 and 12) and ritterazine K.4b However, owing to the complexity of the targets, only very small amounts of these substances were produced and in very low overall yields (2 mg of 1, 10^{-5} %, for instance), 4a not suitable to supply sufficient samples at reasonable cost for extended biological evaluation. However, the recent enantioselective total synthesis of cephalostatin 1 by Shair^{3c} does offer a potentially useful approach to scale-up and represents a splendid contribution. Meanwhile, a number of SAR studies concerned with the cephalostatins were initiated by

various research groups in an effort to discover the minimum pharmacophore required to maintain potent cancer cell growth inhibitory behavior. 5,6

The urgency of the practical syntheses and structural simplification endeavor was elevated when Vollmar and colleagues^{3e,7} began to elucidate the unique mechanism of cephalostatin 1 (1). Because of their detailed research results, it is clear that 1 evokes a new cytochrome c-independent apoptosis signaling pathway. This is in contrast to most of the well-known anticancer drugs, which act in a cytochrome c-dependent route. ${}^{3e,7b-d}$ The lack of cytochrome c release by 1

Received: January 23, 2012 Published: May 18, 2012

indicates that it induces appotosis in cancer cells via caspase-9 activation without formation of an apoptosome (complex of cytochrome c, procaspase-9, and the cytosolic factor Apaf-1). As part of this unique apoptosis mechanism, 1 has been found to selectively release mitochondrial Smac (second mitochondriaderived activator of caspases), necessary for caspase-9 and caspase-2 activations. 7a Furthermore, 1 produces an endoplasmic reticulum stress response, inducing caspase-4, which activates the caspase-9 route to apoptosis. The new pathway is marked by structural changes in the mitochondria. The Current research results^{7a} strongly indicate that the action of 1 on cancer cells is very complex and that further mechanistic investigation will provide many new insights of importance to cancer biology and further preclinical development of the cephalostatins. Shair and colleagues^{3a} have recently reported that cephalostatin 1 (1) is one of a group of molecules that target oxysterol binding protein and its closest paralog and are useful as probes to reveal the functions of these proteins.

In order to extend our SAR studies¹ of **1**, we next focused on examining an extension of the C-17 side chain of a bis-steroidal pyrazine with a 5-substituted α -pyrone system characteristic of the bufadienolides, a large family of broadly bioactive animal and plant steroids originally isolated from toad venom. Certain members of the bufadienolides possess significant antineoplastic behavior,⁸ typified by bufalin (**2**),^{8b-d} and a broad spectrum of other biological activities, including regulation of hypertension. ^{8e,f} Also, in an effort to keep synthetic complexity to a minimum, a symmetrical C-17-pyrone bis-steroidal pyrazine was targeted. Cephalostatin 12^{2f} is the only natural C_2 -symmetric cephalostatin.

■ RESULTS AND DISCUSSION

Initial efforts (Scheme 1)9 were directed at the formation of a simple symmetric pyrazine derived from 3-oxo- 17β -hydroxy- 5α -androstane (3, 5α -dihydrotestosterone) that would be suitable for condensation with the required pyran-2-one synthon, according to the method of Liu and Meinwald.¹⁰ Acylation of 3 afforded 4a, and subsequent conversion of 4a to **5a** employing phenyltrimethylammonium tribromide (PTAB)^{9e} was readily accomplished (98% yield from 3). Transformation of **5a** to **6a** with NaN₃^{9c,d} and catalytic NaI^{9f} in dimethylformamide (DMF) proceeded well (75% yield). Direct dimerization of 6a to give symmetric pyrazine 7a was attempted with p-toluenesulfonic acid (p-TsOH)^{9b,c} as catalyst, but a dehydropyrazine was formed instead. Subsequent attempts at conversion to 7a with O2 and again with 10% Pd/C proved ineffective. Hence, an alternate route was chosen in which 6a was first reduced catalytically with H2 and 10% Pd/C in toluene at 40 psi for 2 h. 10 Subsequent acid-catalyzed dimerization with p-TsOH in EtOH provided 7a in moderate yield (45%), with an overall yield of 32% from 3.

Rather than use 7a as a substrate to obtain the required 17-oxo derivative, it was more practical to repeat the sequence depicted in Scheme 1 with 3,17-dioxo- 5α -androstane (4b) as starting material, which was prepared by oxidation of 3 using pyridinium dichromate. Subsequent bromination to afford 5b, followed by amination (6b) and dimerization, yielded 17,17'-dioxopyrazine 7b in five steps from 3 in an overall yield of 28%. Efforts at transformation of ketone 7b to 16,16'-dienyl-17,17'-ditriflate 8 using lithium diisopropylamide (LDA)¹⁰ as base resulted in decomposition, but the hindered base 2,6-di-tert-butyl-4-methylpyridine proved very effective as a substitute for LDA in promoting the enolization of 7b, which provided 8 in

Scheme 1. Synthesis of 17,17'-Diacetate 7a and 17,17'-Diketone 7b from 3-Oxo-17 β -hydroxy-5 α -androstane (3)

OH Ac₂O, DMAP or PDC

3 4a, R₁ = OAc, R₂ = H, 99% or 4b, R₁, R₂ = O, 99%

5a, R₁ = OAc, R₂ = H, 99%
5b, R₁, R₂ = O, 95%

6a, R₁ = OAc, R₂ = H, 75%
6b, R₁, R₂ = O, 75%

7a, R₁ = OAc, R₂ = H, 45%
7b, R₁, R₂ = O, 45%

7b
$$\frac{(CF_3SO_2)_2O}{2,6-di-t-Bu-4-Me-pyr}$$

8, 84%

good yield (84%) as a substrate for subsequent condensation with a trimethylstannyl derivative.

Adaptation of the sequence detailed by Cho¹¹ for decarboxylation/bromination of coumalic acid (9, Scheme 2) with *N*-bromosuccinimide (NBS) and lithium acetate led to 5-bromo-2*H*-pyran-2-one (10) in low yield (15%), accompanied by a dibrominated byproduct that was obtained in similar yield. By employment of methods analogous to those used to give

Scheme 2. Synthesis of Bis-bufadienolidepyrazine 12

phenylstannates, as described by Liu and Meinwald,¹⁰ treatment of **10** with hexamethylditin and the catalyst Pd(PPh₃)₄ yielded 5-(trimethylstannyl)-2*H*-pyran-2-one (**11**) in good yield (70%). Stille coupling¹² of enol triflate **8** and stannyl pyrone **11** employing Pd(PPh₃)₄ as catalyst resulted in the first synthesis (31% yield) of a bis-bufadienolide pyrazine (**12**), which was obtained from **3** in 8% overall yield.

In order to extend the bis-steroidal pyrazine 17-pyrone SAR probes, the general approach was extended as follows to a dihydro- γ -pyrone. The rationale for a dihydro- α - or dihydro- γ -pyrone unit at or near the steroid C-17 position receives some support from nature. A cancer cell growth inhibitory series of terrestrial plant constituents that contain a dihydro- α -pyrone ring incorporated at the steroid C-22 position, known as the withaphysalins, of which withaphysalin F^{13a} (from *Physalis angulata*) is illustrative, are found in the Solanaceae family. Biologically active γ -pyrone-based natural products have been isolated from plant, fungal, and animal sources (e.g., viridoxin A^{13b} from the fungus *Metarhizium flavoviride* and auripyrone A^{13c} from the mollusk *Dolabella auricularia*), and interest in the properties of the widespread γ -pyrones has increased recently. ^{13d},e

In the next approach to formation of a cephalostatin analogue with a structurally very simple replacement for the E and F rings of 1, that is, a dihydro-γ-pyrone unit, we utilized pyrazine 14 (prepared in six synthetic steps from keto-alcohol 13 as described in our previous report, which was focused on replacement of the E and F rings with a rhamnoside). Selective oxidation (Scheme 3) of 14 with manganese dioxide afforded dialdehyde 15. Condensation of 15 with Danishefsky's diene¹⁴ employing a Diels-Alder-type cycloaddition provided, albeit in low yield (17%), the 2,3-dihydro-4-pyrone 16. The configuration at C-21 on the pyrone rings was deduced from analysis of the ¹H and ¹H-¹H COSY NMR spectra. A pair of doublets at δ 7.38 (J = 6.0 Hz) and 7.35 ($\tilde{J} = 6.0$ Hz), which were assigned to the C-22 protons of the rings, were coupled with the signal at δ 5.42 (I = 6.0 Hz, H-23). The multiplet at δ 5.20 (H-20) shows a vicinal coupling of 8.5 Hz and an allylic coupling of 2.5 Hz, corresponding to a strong correlation with the multiplet at δ 4.95–5.03 (H-21) and a minor correlation with the signal at δ 2.53–2.68 (H-25). These data are consistent with a structure that includes both (R)- and (S)pyrone substituents, resulting from both exo and endo hetero-Diels-Alder cycloaddition, which gives rise to complex

Scheme 3. Synthesis of Dihydro-4-pyrone 16 from 13

multiplets for each of the C-20 and C-21 protons and a pair of doublets for the two C-22 protons. The signals from the C-23 protons are coincident, and those from H-25 are buried in the steroid skeleton signals.

Compounds 7a,b, 8, 12, and 16 were screened for cancer cell growth inhibitory activity in the murine P388 lymphocytic leukemia cancer cell line (Table 1). The steroidal pyrazine side-

Table 1. Murine P388 Lymphocytic Leukemia Cell Line Results (ED₅₀ values)

μ g/mL	$\mu\mathrm{M}$
$10^{-7} - 10^{-9}$	$10^{-7} - 10^{-9}$
0.072	0.076
60	91
68	120
35	42
41	57
20	25
	10 ⁻⁷ -10 ⁻⁹ 0.072 60 68 35 41

chain pyrones 12 (ED $_{50}$ 41 $\mu g/mL$, 57 μM) and 16 (ED $_{50}$ 20 $\mu g/mL$, 25 μM) did not exhibit significant (ED $_{50} \leq$ 10 $\mu g/mL$) activity. However, these SAR results added further confirmation of Winterfeldt's carry analysis of cephalostatin 1 structural requirements for strong cancer cell growth inhibitory activity, particularly with respect to the need for a C-14 double bond, C-12 oxygenation (alcohol or ketone), and a 17 α -hydroxy group for optimum activity. Presumably, there are a number of other less obvious molecular features of the cephalostatin-type steroids that are critical to their quite unique potency against cancer cell growth and mechanisms of biological activity. The uncovering of such structural features is a goal of future SAR investigations of the cephalostatins. Meanwhile, we are further evaluating the bis-steroidal pyrazine pyrones described herein for other biological activities.

■ EXPERIMENTAL SECTION

General Experimental Procedures. Melting points are uncorrected and were determined with an Electrothermal 9100 apparatus. The IR spectra were obtained using a Thermo-Nicolet (Thermo Fisher Scientific) Avatar 360 Series FT-IR spectrometer. The ¹H NMR and ¹³C NMR spectra were recorded employing Varian Gemini 300 and Varian Unity 500 instruments using CDCl₃ (TMS internal reference) as solvent unless otherwise noted; bs refers to broad singlet. High-resolution APCI⁺ (atmospheric pressure chemical ionization) mass spectra were obtained with a Jeol JMS-LCmate mass spectrometer. Elemental analyses were determined by Galbraith Laboratories, Inc.

Ether refers to diethyl ether, TEA to triethylamine, DCM to dichloromethane, DMAP to 4-dimethylaminopyridine, Ar to argon gas, and rt to room temperature. All solvents were redistilled. All chemicals were purchased from either Sigma-Aldrich Corp. or Acros Organics (Thermo Fisher Scientific). Reactions were monitored by thin-layer chromatography (TLC) using Analtech silica gel GHLF Uniplates and were visualized with either phosphomolybdic acid (10 wt %/wt solution in EtOH) or iodine. Solvent extracts of aqueous solutions were dried over anhydrous magnesium or sodium sulfate. Where appropriate, the crude products were purified by column chromatography (CC) on silica gel (70–230 mesh ASTM) from E. Merck.

3-Oxo-17 β -acetoxy-5 α -androstane (4a). To a stirred solution of dihydrotestosterone (3, 1.0 g, 3.45 mmol) and DMAP (cat) in pyridine (10 mL) at rt under Ar was added acetic anhydride (10 mL, 106 mmol) dropwise, and stirring continued for 6 h. The mixture was cooled to 0 °C, H₂O (15 mL) and ether (15 mL) were added, and the phases were separated. After extraction of the aqueous phase with ether (2 × 15 mL), the combined organic phase was washed with HCl (1 M, 2 \times 15 mL), NaHCO₃ (saturated aqueous, 2 \times 15 mL), and H₂O (15 mL). After drying, concentration of the organic phase afforded a colorless, amorphous solid. Crystallization from cyclohexane—acetone provided 4a as colorless crystals (1.14 g, 99%): mp 155–156 °C [lit. 15 mp 157–158.5 °C]; ¹H NMR (300 MHz, CDCl₃) δ 4.60 (1H, t, J = 8.8 Hz, H-17 α), 2.42 (1H, m, H-2 α), 2.32 (1H, t, J = 9.0 Hz, H-2 β), 2.21 (1H, t, J = 8.8 Hz, H-4 α), 2.10 (1H, m, H-16 β), $2.04 (1H, m, H-4\beta), 2.03 (3H, s, -OCOCH₃), 1.75-1.05 (14H), 1.02$ (3H, s, CH₃), 0.94 (1H, m), 0.81 (3H, s, CH₃), 0.76 (1H, m); ¹³C NMR (126 MHz, CDCl₃) δ 211.8, 171.2, 82.7, 53.7, 50.6, 49.6, 44.6, 42.6, 38.5, 38.1, 36.8, 35.7, 35.2, 31.2, 28.8, 27.5, 23.5, 21.1, 20.9, 12.1, 11.4; HRMS (APCI⁺) m/z 333.24144 [M + H]⁺ (calcd for $C_{21}H_{33}O_{3}$, 333.24297); anal. C 75.76, H 9.89%, calcd for $C_{21}H_{32}O_3$, C 75.86, H

 2α -Bromo-3-oxo-17 β -acetoxy-5 α -androstane (**5a**). To a stirred solution of 4a (0.50 g, 1.50 mmol) in THF (5 mL) at 0 °C under Ar was added phenyltrimethylammonium tribromide (0.59 g, 1.56 mmol), and the solution immediately turned orange. After 20 min stirring, a tan precipitate formed, which dissolved upon addition of H_2O (5 mL). After extraction with DCM (2 × 5 mL) the combined organic phase was concentrated in vacuo, dissolved in minimal DCM, and subjected to CC (3:7 EtOAc-n-hexane) to yield a light tan, amorphous solid. Crystallization from cyclohexane-acetone gave 5a as colorless crystals (0.61 g, 99%): mp 171.8–172.4 °C; IR (neat) $\nu_{\rm max}$ 1733 (C=O, ketone and ester) cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 4.73 (1H, q, J = 8.8 Hz, H-2 β), 4.58 (1H, t, J = 8.8 Hz, H-17 α), 2.64 (1H, q, J = 8.8 Hz, H-1 α), 2.42 (2H, m, H-4 α , β), 2.17 (1H, m, H- 16β), 2.03 (3H, s, OCOCH₃), 1.80 (1H), 1.78–1.20 (14H), 1.08 (3H, s, CH₃), 0.98 (1H, m), 0.85 (1H, m), 0.79 (3H, s, CH₃); ¹³C NMR (126 MHz, CDCl₃) δ 200.9, 171.1, 82.5, 54.3, 53.5, 51.6, 50.4, 47.4, 43.8, 42.6, 39.0, 36.6, 34.7, 31.0, 28.2, 27.5, 23.5, 21.1, 21.0, 12.1; HRMS (APCI⁺) m/z 411.15272 [M + H]⁺ (calcd for $C_{21}H_{32}^{79}BrO_{34}$ 411.15348); anal. C 61.42, H 7.56%, calcd for C₂₁H₃₁BrO₃, C 61.37, H 7.60%

2-Amino-3-oxo-17 β -acetoxy-5 α -androstan-1-ene (**6a**). To a stirred solution of **5a** (0.50 g, 1.22 mmol) in DMF (10 mL) at rt under Ar were added NaN $_3$ (1.00 g, 15.4 mmol) and NaI (cat). The suspension was heated to 60 °C for 2 h and then cooled to rt, H $_2$ O (10

mL) was added, and the solution was extracted with ether (5 × 10 mL). The combined organic phase was concentrated in vacuo, and separation (CC, 1% TEA in 7:3 n-hexane—EtOAc) yielded a yellow, amorphous solid. Crystallization from cyclohexane—acetone gave **6a** as light yellow needles (0.46 g, 75%): mp 149.8–150.5 °C; IR (neat) $\nu_{\rm max}$ 1739, 1715 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.09 (1H, s, H-1), 4.61 (1H, t, J = 8.8 Hz, H-17 α), 3.43 (2H, bs, NH₂), 2.41 (1H, t, J = 8.8 Hz, H-4 α), 2.26 (1H, dd, J = 8.8 Hz, H-4 β), 2.16 (1H, m, H-16 β), 2.04 (3H, s, OCOCH₃), 1.91 (1H, m, H-5 α), 1.80–1.20 (14H), 1.08 (3H, s, CH₃), 0.98 (1H, m), 0.85 (1H, m), 0.79 (3H, s, CH₃); ¹³C NMR (126 MHz, CDCl₃) δ 195.7, 171.7, 137.7, 126.5, 82.6, 51.2, 50.7, 44.8, 42.8, 40.2 (C-4), 38.1, 36.8, 35.3, 30.9, 27.5, 27.2, 27.1, 23.4, 21.1, 20.9, 13.9, 12.2; HRMS (APCI⁺) m/z 346.23829 [M + H]⁺ (calcd for C₂₁H₃₂NO₃, 346.23822); anal. C 73.32, H 9.11%, calcd for C₂₁H₃₁NO₃, C 73.01, H 9.04%.

Bis(17 β -acetoxy-5 α -androstan[3,2-e,3',2'])pyrazine (**7a**). To a stirred solution of amine 6a (0.50 g, 1.66 mmol) in toluene (10 mL) was added 10% Pd/C (0.28 g, 15 mol % Pd) in a heavy-walled hydrogenation flask. The flask was purged with H₂ (4×) before hydrogenation at 40 psi for 2 h. The black suspension was collected by filtration and washed with DCM (10 mL), and concentration of the organic phase in vacuo provided the 2α -amino ketone^{3b,9a,b} intermediate as a brown solid, which was immediately used for dimerization without further purification. The 2α -amino ketone was dissolved in EtOH (5 mL), and pTsOH (cat) was added. After stirring for 48 h, the reaction solution was filtered through a pad of silica gel and washed with EtOAc (100 mL). The organic phase was concentrated in vacuo, and fractionation (CC, 1% TEA in 7:3 nhexane-EtOAc) yielded an amorphous solid. Cystallization from cyclohexane-acetone provided 7a as colorless needles (0.24 g, 45%): mp 240 °C (dec); IR (neat) $\nu_{\rm max}$ 1738, 1717 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 4.61 (1H, t, J = 8.8 Hz, H-17 α), 2.91 (1H, d, J = 8.8 Hz, H-1 α), 2.71 (1H, dd, H-4 α), 2.55 (2H, m, H-1 β , H-4 β), 2.16 (1H, m, H-16 β), 2.04 (3H, s, OCOCH₃), 1.81–1.20 (14H), 1.08 (3H, s, CH₃), 0.98 (1H, m), 0.85 (1H, m), 0.79 (3H, s, CH₃); ¹³C NMR (126 MHz, CDCl₃) δ 171.2, 133.6, 127.1, 82.7, 52.0, 50.8, 50.6, 43.5, 42.8, 42.5, 41.7, 39.9, 38.3, 36.9, 35.6, 35.4, 30.9, 27.5, 27.0, 23.5, 21.2, 21.1; HRMS (APCI⁺) m/z 657.46041 [M + H]⁺ (calcd for $C_{42}H_{61}N_2O_4$, 657.46313); anal. C 76.94, H 9.11, N 4.31%, calcd for C₄₂H₆₀N₂O₄, C 76.79, H 9.21, N 4.26%.

3,17-Dioxo-5 α -androstane (4b). To a stirred solution of dihydrotestosterone (3, 1.0 g, 3.44 mmol) in DCM (10 mL) at rt under Ar was added pyridinium dichromate (2.10 g, 5.58 mmol). After stirring for 2 h, the black suspension was collected by vacuum filtration through a short pad of silica gel with EtOAc (100 mL) as eluent. The organic extract was concentrated in vacuo, and purification (CC, 3:2 nhexane-EtOAc) led to a colorless, amorphous solid. Crystallization from cyclohexane-acetone provided 4b as colorless crystals (0.98 g, 99%): mp 134.1–134.5 °C [lit. 16 mp 133.5–134.0 °C]; IR (neat) ν_{max} 1742, 1739 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 2.42 (1H, m, H-2 α), 2.32 (1H, t, J = 9.0 Hz, H-2 β), 2.21 (1H, t, J = 8.8 Hz, H-4 α), 2.10 $(1H, m, H-16\beta)$, 2.04 $(1H, m, H-4\beta)$, 1.82 $(1H, m, H-16\alpha)$, 1.76–1.11 (14H), 1.03 (3H, s, CH₃), 0.94 (1H, m), 0.88 (3H, s, CH₃), 0.76 (1H, m); 13 C NMR (126 MHz, CDCl₃) δ 220.9 (C-17), 210.9 (C-3), 53.4, 50.7, 47.2, 46.1, 44.0, 37.9, 37.5, 35.3, 34.5, 31.0, 30.0, 28.1, 21.2, 20.2, 13.3, 10.9; HRMS (APCI⁺) m/z 289.2165 [M + H]⁺ (calcd for $C_{19}H_{29}O_2$, 289.2168); anal. C 79.28, H 9.95%, calcd for $C_{19}H_{28}O_2$, C 79.12, H 9.78%.

2α-Bromo-3,17-dioxo-5α-androstane (5b). To a stirred solution of 4b (1.5 g, 5.20 mmol) in THF (15 mL) at 0 °C under Ar was added PTAB (2.20 g, 5.85 mmol). The solution immediately turned orange and was stirred for 20 min. A tan precipitate formed that dissolved on addition of H₂O (10 mL). After extraction with DCM (2 × 15 mL), the combined organic phase was concentrated in vacuo, and purification (CC, 7:3 *n*-hexane–EtOAc) yielded a light tan, amorphous solid. Crystallization from cyclohexane–acetone gave 5b as colorless crystals (1.81 g, 95%): mp 136.4–136.7 °C; IR (neat) ν_{max} 1733, 1716 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 4.73 (1H, q, J = 8.8 Hz, H-2 β), 2.65 (1H, q, J = 8.8 Hz, H-1 β), 2.45 (3H, m, H-4 α , H–16 β , H-16 α), 2.11 (1H, t, H-1 α), 1.96 (1H), 1.89–1.18 (14H), 1.11

(3H, s, CH₃), 0.94 (1H, m), 0.88 (3H, s, CH₃), 0.76 (1H, m); 13 C NMR (126 MHz, CDCl₃) δ 220.9 (C-17), 211.1 (C-3), 54.3 (C-2), 53.5, 53.2, 51.0, 50.5, 47.1, 46.9, 43.3, 35.2, 34.0, 30.1, 29.8, 27.6, 21.2, 20.3, 17.2, 13.3, 11.6; HRMS (APCI⁺) m/z 367.1321 [M + H]⁺ (calcd for C₁₉H₂₈⁷⁹BrO₂ 367.1273); anal. C 62.21, H 7.39%, calcd for C₁₉H₂₇BrO₂, C 62.13, H 7.41%.

2-Amino-3,17-dioxo-5 α -androstan-1-ene (**6b**). To a stirred solution of 5b (1.5 g, 4.08 mmol) in DMF (15 mL) at rt under Ar were added NaN₃ (2.70 g, 41.5 mmol) and NaI (cat), and the suspension was heated to 60 °C for 2 h. After cooling to rt, the mixture was diluted with H_2O (15 mL) and extracted with ether (5 × 20 mL). The combined organic phase was concentrated in vacuo, and the residue was purified (CC, 1% TEA in 7:3 n-hexane-EtOAc) to yield a yellow, amorphous solid. Crystallization from cyclohexane-acetone gave **6b** as light yellow needles (0.92 g, 75%): mp 135.3-175.9 °C; IR (neat) $\nu_{\rm max}$ 1734, 1700 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 6.07 (1H, s, H-1), 3.45 (2H, bs, NH₂), 2.45 (3H, m, H-4 α , H-16 β , H-16 α), 2.11 (1H, H-16 β), 1.94–1.16 (14H), 1.00 (3H, s, CH₃), 0.94 (1H, m), 0.88 (3H, s, CH₃), 0.76 (1H, m); ¹³C NMR (126 MHz, CDCl₃) δ 220.1 (C-17), 195.0 (C-3), 137.4 (C-1), 125.4 (C-2), 50.9, 50.8, 47.3, 44.3, 39.6, 37.7, 35.3, 34.6, 31.0, 29.7, 26.6, 21.1, 20.2, 13.4, 11.9; HRMS (APCI⁺) m/z 302.2120 [M + H]⁺ (calcd for C₁₉H₂₈NO₂, 302.2120); anal. C 75.42, H 9.13, N 4.56%, calcd for C₁₉H₂₇NO₂, C 75.71, H 9.03, N 4.65%.

Bis(17-oxo-5 α -androstan[2,3-b:2',3'-e])pyrazine (**7b**). To a stirred solution of 6b (0.50 g, 1.66 mmol) in toluene (10 mL) was added 10% Pd/C (0.28 g, 15 mol % Pd) in a heavy-walled hydrogenation flask. The flask was purged with H₂ (4×) before hydrogenation at 40 psi for 2 h. The black suspension was then removed by filtration and washed with DCM (10 mL). The combined organic extract was concentrated in vacuo to yield the 2α -amino ketone intermediate as a brown solid, which was immediately used without further purification for dimerization. The amine (1.66 mmol) was dissolved in EtOH (5 mL), and pTsOH (cat) added. After stirring for 48 h, the reaction solution was filtered (vacuum) through a short pad of silica gel and washed with EtOAc (100 mL). The solvent extract was concentrated in vacuo, adsorbed onto silica gel, and subjected to CC (7:3 n-hexane-EtOAc, 1% TEA) to give an amorphous solid. Crystallization from cyclohexane-acetone provided 7b as colorless needles (0.19 g, 45%): mp 235 °C (dec); IR (neat) $\nu_{\rm max}$ 1737 cm $^{-1}$; $^{1}{\rm H}$ NMR (300 MHz, $CDCl_3$) δ 2.93 (1H, d, J = 8.8 Hz, H-1 α), 2.68 (1H, dd, H-4 α), 2.50 $(3H, m, H-1\beta, H-4\beta, H-16\beta), 2.19 (1H, m, H-16\alpha), 1.88-0.96 (16H),$ 0.88 (3H, s, CH₃), 0.85 (1H, m), 0.82 (3H, s, CH₃); ¹³C NMR (126 MHz, CDCl₃) δ 220.4 (C-17), 133.6 (C-2), 127.1 (C-3), 53.3, 50.8, 47.1, 45.4, 41.9, 41.3, 35.3, 34.9, 34.4, 33.3, 32.6, 31.1, 29.9, 27.6, 24.2, 21.3, 20.0, 13.2, 11.5; HRMS (APCI⁺) m/z 569.4158 [M + H]⁺ (calcd for C₃₈H₅₃N₂O₂, 569.4107); anal. C 80.31, H 9.11, N 4.86%, calcd for $C_{38}H_{52}N_2O_{27}$ C 80.24, H 9.21, N 4.92%. Bis(17-triflyloxy- Δ^{16} -5 α -androstan[2,3-b:2',3'-e])pyrazine (8). To

Bis(17-triflyloxy- Δ^{16} -5α-androstan[2,3-b:2',3'-e])pyrazine (8). To a stirred solution of 7b (0.40 g, 0.70 mmol) and 2,6-di-tert-butyl-4-methylpyridine (0.44 g, 2.14 mmol) in DCM (4 mL) at -10 °C under Ar was added (dropwise) triflic anhydride (0.36 mL, 2.14 mmol). After 18 h, H₂O (8 mL) and DCM (4 mL) were added, and the aqueous phase was extracted with DCM (2 × 8 mL). The combined organic phase was concentrated in vacuo, and silica gel separation (CC, 1:1 *n*-hexane–EtOAc) yielded 8 as an amorphous, light yellow solid (0.49 g, 84%): mp 190 °C (dec); ¹H NMR (300 MHz, CDCl₃) δ 5.77 (1H, m, H-16), 2.91 (1H, d, J = 8.8 Hz, H-1α), 2.71 (1H, dd, H-4α), 2.50 (2H, m, H-1β, H-4β), 1.81–1.11 (16H), 1.00 (3H, s, CH₃), 0.94 (1H, m), 0.84 (3H, s, CH₃); ¹³C NMR (126 MHz, CDCl₃) δ 158.7 (C-17), 152.0 (OSO₂CF₃), 135.6 (C-2), 129.1 (C-3), 113.9 (C-16), 53.6, 53.5, 45.1, 44.3, 41.4, 35.3, 34.9, 32.8, 32.2, 29.9, 29.6, 28.0, 27.5, 19.9, 14.7, 11.4; HRMS (APCI⁺) m/z 833.3094 [M + H]⁺ (calcd for C₄₀H₅₁F₆N₂O₆S₂, 833.3093).

5-Bromo-2H-pyran-2-one (10). To a stirred solution of coumalic acid (9, 10.0 g, 71.4 mmol) in CH_3CN-H_2O (9:1; 400 mL) at rt were added LiOAc (5.60 g, 84.9 mmol), NBS (19.0 g, 107 mmol), and Bu₄NI (1.0 g, 4 mol %). After the mixture was stirred for 10 days, H_2O (250 mL) and DCM (250 mL) were added, and the aqueous phase was extracted with DCM (2 × 100 mL). The combined solvent extract

was concentrated in vacuo, and separation of the residue (CC, 9:1 n-hexane–EtOAc) led to **10** as a light tan, crystalline solid (1.9 g, 15%); mp 39.4–39.9 °C; IR (neat) $\nu_{\rm max}$ 1738 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 7.63 (1H, dd, J = 2.7, 1.2 Hz, H-6), 7.32 (1H, dd, J = 9.9, 2.7 Hz, H-4), 6.28 (1H, dd, J = 1.2, 9.9 Hz, H-3); ¹³C NMR (CDCl₃, 126 MHz) δ 171.0 (C-2), 146.7 (C-4), 133.8 (C-6), 120.9 (C-3), 100.6 (C-5).

5-(Trimethylstannyl)-2H-pyran-2-one (11) (ref 10). To a stirred solution of pyrone 10 (0.70 g, 4.00 mmol) in THF (7 mL) at rt under Ar were added Pd(PPh₃)₄ (0.20 g, 0.173 mmol) and hexamethylditin (5.0 g, 15.3 mmol), and the solution was heated to reflux for three days. The black mixture was cooled to rt and concentrated in vacuo. Separation of the residue (CC, 3:17 EtOAc–n-hexane) yielded 11 as a colorless oil (0.73 g, 70%): IR (neat) $\nu_{\rm max}$ 1736 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.23 (1H, dd, J = 2.4, 9.3 Hz, H-4), 7.17 (1H, dd, J = 1.8, 2.4 Hz, H-6), 6.26 (1H, dd, J = 1.8, 9.3 Hz, H-3), 0.25 (9H, s, (CH₃)₃); ¹³C NMR (126 MHz, CDCl₃) δ 171.0 (C-2), 146.7 (C-4), 134.6 (C-6), 120.9 (C-3), 111.8 (C-5), -2.1 (3 × -CH₃).

Bis(5α -bufa-16,20(21),22-trienolide-[2,3-b:2',3'-e])pyrazine (12). To a stirred solution of triflate 8 (0.90 g, 1.08 mmol) and pyrone 11 (0.64 g, 2.45 mmol) in THF (10 mL at rt under Ar) were added LiCl (0.68 g, 16.0 mmol) and Pd(PPh₃)₄ (0.220 g, 0.190 mmol), and the mixture was heated under reflux for three days. After the solution had cooled to rt, brine (10 mL) and DCM (10 mL) were added successively, and the aqueous phase was extracted with DCM (3 × 10 mL). The combined organic phase was concentrated in vacuo, and the residue was separated (CC, 1:20 acetone-n-hexane) to afford 12 as an amorphous solid. Crystallization from cyclohexane-acetone provided bis-bufadienolide 12 as colorless needles (0.24 g, 31%): mp 242 °C (dec); IR (neat) $\nu_{\rm max}$ 1722 cm⁻¹; ¹H NMR (126 MHz, CDCl₃) δ 7.25 (1H, dd, J = 2.4, 9.3 Hz, H-22), 7.19 (1H, dd, J = 1.8, 2.4 Hz, H-21),6.46 (1H, dd, *J* = 1.8, 9.3 Hz, H-23), 5.87 (1H, m, H-16), 2.92 (1H, d, $I = 8.8 \text{ Hz}, \text{ H-}1\alpha$), 2.70 (1H, dd, H- 4α), 2.49 (2H, m, H- 1β , H- 4β), 1.84-1.11 (16H), 0.95 (3H, s, CH₃), 0.91 (1H, m), 0.85 (3H, s, CH₃); 13 C NMR (126 MHz, CDCl₃) δ 171.1, 146.9, 145.3, 135.1, 129.7, 127.7, 123.8, 120.9, 113.1, 53.9, 53.1, 45.5, 44.1, 41.9, 35.7, 34.4, 32.5, 32.1, 29.6, 29.1, 28.6, 27.1, 19.1, 14.2, 11.5; HRMS (APCI⁺) m/z725.4317 $[M + H]^+$ (calcd for $C_{48}H_{57}N_2O_4$, 725.4318); anal. C 79.36, H 7.92, N 3.79%, calcd for C₄₈H₅₆N₂O₄, C 79.52, H 7.79, N 3.86%.

Aldehyde 15. To a solution of alcohol 14¹ (120 mg, 0.183 mmol) in 2-propanol (10 mL) was added $\mathrm{MnO_2}$ (0.4 g, 4.58 mmol), and the mixture was stirred at rt. After 24 h the mixture was filtered through Celite, which was washed with DCM. Removal of solvent from the combined filtrate yielded the crude product as a solid. Purification (CC, 30% DCM-MeOH) and crystallization from DCM-EtOAc yielded **15** (80 mg, 0.122 mmol, 66%): mp 250 °C (dec); $[\alpha]^{25}_{D}$ +29.8 (c 0.48, DCM); IR (film) $\nu_{\rm max}$ 3368, 3275, 2915, 1666, 1400, 1153 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 10.11 (2H, d, J = 8.7 Hz, H-21), 5.80 (2H, d, J = 8.7 Hz, H-20), 4.57 (2H, s, H-11), 3.13 (2H, d, J = 16.2 Hz, H-1 β), 2.86 (2H, dd, I = 5.6, 18.2 Hz, H-4 α), 2.74–2.58 (12H, m), 1.35 (6H, s), 1.13 (6H, s), 2.05–0.94 (22H, m); 13 C NMR (125 MHz, CDCl₃) δ 190.7 (C-21), 179.1 (C-17), 148.6 (C-2/3), 148.5 (C-2/3), 123.4 (C-20), 67.8 (C-11), 56.94, 56.87, 48.1, 46.6, 45.0, 42.2, 35.9, 34.8, 33.1, 32.0, 30.7, 27.8, 23.8, 21.5, 14.4; HRMS (APCI⁺) m/z 653.4236 [M + H]⁺ (calcd for C₄₂H₅₇N₂O₄, 653.4318); anal. C 70.99, H 9.06, N 3.89%, calcd for C₄₂H₅₆N₂O₄·4CH₃OH, C 70.74, H 9.29, N 3.59%.

Bis-dihydro-4-pyrone 16. To a solution of aldehyde 15 (15 mg, 0.023 mmol) and Danishefsky's diene (10 μ L, 0.051 mmol)¹⁴ at -78 °C in DCM (10 mL) under Ar was added BF₃·Et₂O (12 μ L, 0.012 mmol). The reaction mixture was stirred for 2 h at -78 °C and then allowed to equilibrate to rt. After 48 h, no change was detectable by TLC, and an additional 12 μ L of BF₃·Et₂O was added. The reaction mixture was stirred at rt for 14 days and then quenched with NaHCO₃ (saturated aqueous) and extracted with EtOAc. Purification (CC, 7:3 acetone–DCM) led to pyrazine 16 (3 mg, 17%): [α]²⁵_D +106.2 (α 0.10, DCM); IR (neat) ν max 3363 (OH), 1672, 1401 (pyrazine ring) cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.38 (1H, d, J = 6.0 Hz, H-22), 7.35 (1H, d, J = 6.0 Hz, H-22), 5.42 (2H, dd, J = 1.0, 6.5 Hz, H-23), 5.20 (2H, m, J = 2.5, 8.5 Hz, H-20), 5.03–4.95 (2H, m, H-21), 4.55

(2H, s, H-11), 3.13 (2H, d, J = 16 Hz, H-1 β), 2.82 (2H, bd, J = 17 Hz), 2.68–2.53 (4H, m, H-25), 2.44–2.36 (2H, m), 2.31–2.21 (2H, m), 2.06–1.83 (6H, m), 1.68–1.42 (8H, m), 1.14–1.05 (10H, m), 0.98–0.83 (4H, m); 13 C NMR (125 MHz, CDCl₃) δ 192.3 (C-24), 163.4 (C-22), 148.7 (C-2/3 or C-17), 148.4 (C-2/3 or C-17), 119.2 (C-20), 106.9 (C-23), 68.3, 67.9, 57.6, 54.6, 45.6, 45.1, 44.2, 42.5, 36.0, 34.9, 32.1, 31.0, 29.7, 27.9, 27.4, 24.2, 20.4, 14.4; HRMS (APCI⁺) m/z 789.4840 [M + H]⁺ (calcd for C₅₀H₆₅N₂O₆, 789.4843).

ASSOCIATED CONTENT

Supporting Information

¹H and ¹³C NMR spectra and a COSY spectrum of compound **16**.¹⁷ This material is available free of charge via the Internet at http://pubs.acs.org.

AUTHOR INFORMATION

Corresponding Author

*Tel: (480) 965-3351. Fax: (480) 965-2747. E-mail: bpettit@asu.edu.

Present Address

[†]United States Department of Agriculture, Agricultural Research Service, National Center for Agricultural Utilization Research, Peoria, Illinois 61604, United States.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

The very necessary financial assistance was provided by grants RO1CA90441-02-05 and 5RO1CA90441-07-08 from the Division of Cancer Treatment Diagnosis and Centers, National Cancer Institute, DHHS; the Arizona Biomedical Research Commission; the Robert B. Dalton Endowment Fund; Dr. Alec D. Keith, the J. W. Kieckhefer Foundation; the Margaret T. Morris Foundation; and the Fannie E. Rippel Foundation. For other helpful assistance we thank Drs. J.-C. Chapuis and V. J. R. V. Mukku, as well as F. Craciunescu and M. Dodson.

REFERENCES

- (1) Antineoplastic Agents 563; for series part 562 see: Pettit, G. R.; Mendonça, R. F.; Knight, J. C.; Pettit, R. K. J. Nat. Prod. 2011, 74, 1922—1920
- (2) (a) Pettit, G. R.; Inoue, M.; Kamano, Y.; Herald, D. L.; Arm, C.; Dufresne, C.; Christie, N. D.; Schmidt, J. M.; Doubek, D. L.; Krupa, T. S. J. Am. Chem. Soc. 1988, 110, 2006-2007. (b) Pettit, G. R.; Inoue, M.; Kamano, Y.; Dufresne, C.; Christie, N.; Niven, M. L.; Herald, D. L. J. Chem. Soc., Chem. Commun. 1988, 865-867. (c) Pettit, G. R.; Kamano, Y.; Dufresne, C.; Inoue, M.; Christie, N.; Schmidt, J. M.; Doubek, D. L. Can. J. Chem. 1989, 67, 1509-1513. (d) Pettit, G. R.; Kamano, Y.; Inoue, M.; Dufresne, C.; Boyd, M. R.; Herald, D. L.; Schmidt, J. M.; Doubek, D. L.; Christie, N. D. J. Org. Chem. 1992, 57, 429-431. (e) Pettit, G. R.; Xu, J.-P.; Williams, M. D.; Christie, N. D.; Doubek, D. L.; Schmidt, J. M.; Boyd, M. R. J. Nat. Prod. 1994, 57, 52-63. (f) Pettit, G. R.; Ichihara, Y.; Xu, J.; Boyd, M. R.; Williams, M. D. Bioorg. Med. Chem. Lett. 1994, 4, 1507-1512. (g) Pettit, G. R.; Xu, J. P.; Ichihara, Y.; Williams, M. D. Can. J. Chem. 1994, 72, 2260-2267. (h) Pettit, G. R.; Xu, J.-P.; Schmidt, J. M.; Boyd, M. R. Bioorg. Med. Chem. Lett. 1995, 5, 2027-2032. (i) Pettit, G. R.; Tan, R.; Xu, J. P.; Ichihara, Y.; Williams, M. D.; Boyd, M. R. J. Nat. Prod. 1998, 61, 955-
- (3) (a) Burgett, A. W. G.; Poulsen, T. B.; Wangkanont, K.; Anderson, D. R.; Kikuchi, C.; Shimada, K.; Okubo, S.; Fortner, K. C.; Mimaki, Y.; Koruda, M.; Murphy, J. P.; Schwalb, D. J.; Petrella, E. C.; Cornella-Taracido, I.; Schirle, M.; Tallarico, J. A.; Shair, M. D. *Nat. Chem. Biol.* 2011, 7, 639–647. (b) Kumar, A. K.; La Clair, J. J.; Fuchs, P. L. *Org. Lett.* 2011, 13, 5334–5337. (c) Fortner, K. C.; Kato, D.; Tanaka, Y.;

Shair, M. D. J. Am. Chem. Soc. 2010, 132, 275–280. (d) Lee, S.; LaCour, T. G.; Fuchs, P. L. Chem. Rev. 2009, 109, 2275–2314. (e) Rudy, A.; López-Antón, N.; Dirsch, V. M.; Vollmar, A. M. J. Nat. Prod. 2008, 71, 482–486. (f) Moser, B. R. J. Nat. Prod. 2008, 71, 487–491. (g) Flessner, T.; Jautelat, R.; Scholz, U.; Winterfeldt, E. In Progress in the Chemistry of Organic Natural Products; Herz, W., Falk, H., Kirby, G. W., Eds.; Springer: Vienna, 2004; Vol. 87, pp 1–80.

- (4) (a) LaCour, T. G.; Guo, C.; Bhandaru, S.; Boyd, M. R; Fuchs, P. L. J. Am. Chem. Soc. 1998, 120, 692–707. (b) Jeong, J. U.; Sutton, S. C.; Kim, S.; Fuchs, P. L. J. Am. Chem. Soc. 1995, 117, 10157–10158. (c) Guo, C.; LaCour, T. G.; Fuchs, P. L. Bioorg. Med. Chem. Lett. 1999, 9, 419–424.
- (5) (a) Poza, J. J.; Rodríguez, J.; Jiménez, C. Bioorg. Med. Chem. 2010, 18, 58-63. (b) Nawasreh, M. M. Curr. Org. Chem. 2009, 13, 407-420.
 (c) Lee, S.; Jamieson, D.; Fuchs, F. L. Org. Lett. 2009, 11, 5-8.
 (d) Gryszkiewicz-Wojtkielewicz, A.; Jastrzebska, I.; Morzycki, J. W.; Romanowska, D. B. Curr. Org. Chem. 2003, 7, 1257-1277.
- (6) (a) Taber, D. F.; Joerger, J.-M. J. Org. Chem. 2008, 73, 4155–4159. (b) Nawasreh, M. Bioorg. Med. Chem. 2008, 16, 255–265. (c) Nawasreh, M.; Winterfeldt, E. Curr. Org. Chem. 2003, 7, 649–658. (d) Bäsler, S.; Brunck, A.; Jautelat, R.; Winterfeldt, E. Helv. Chim. Acta 2000, 83, 1854–1880. (e) Drögemüller, M.; Jautelat, R.; Winterfeldt, E. Angew. Chem., Int. Ed. 1996, 35, 1572–1574. (f) Kramer, A.; Ullmann, U.; Winterfeldt, E. J. Chem. Soc., Perkin Trans. 1 1993, 2865–2867.
- (7) (a) Rudy, A.; López-Antón, N.; Barth, N.; Pettit, G. R.; Dirsch, V. M.; Schulze-Osthoff, K.; Rehm, M.; Prehn, J. H. M.; Vogler, M.; Fulda, S.; Vollmar, A. M. *Cell Death Differ.* **2008**, *15*, 1930–1940. (b) Dirsch, V. M.; Müller, I. M.; Eichhorst, S. T.; Pettit, G. R.; Kamano, Y.; Inoue, M.; Xu, J.-P.; Ichihara, Y.; Wanner, G.; Vollmar, A. M. *Cancer Res.* **2003**, *63*, 8869–8876. (c) Müller, I. M.; Dirsch, V. M.; Rudy, A.; López-Antón, N.; Pettit, G. R.; Vollmar, A. M. *Mol. Pharmacol.* **2005**, *67*, 1684–1689. (d) López-Antón, N.; Rudy, A.; Barth, N.; Schmitz, L. M.; Pettit, G. R.; Schulze-Osthoff, K.; Dirsch, V. M.; Vollmar, A. M. *J. Biol. Chem.* **2006**, *281*, 33078–33086.
- (8) (a) Kuo, P.-C.; Kuo, T.-H.; Su, C.-R.; Liou, M.-J.; Wu, T.-S. Tetrahedron 2008, 64, 3392-3396. (b) Kamano, Y.; Yamashita, A.; Nogawa, T.; Morita, H.; Takeya, K.; Itokawa, H.; Segawa, T.; Yukita, A.; Saito, K.; Katsuyama, M.; Pettit, G. R. J. Med. Chem. 2002, 45, 5440-5447. (c) Kamano, Y.; Nogawa, T.; Yamashita, A.; Hayashi, M.; Inoue, M.; Drašar, P.; Pettit, G. R. J. Nat. Prod. 2002, 65, 1001-1005. (d) Kamano, Y.; Yamashita, A.; Takeuchi, H.; Kohyama, K.; Nogawa, T.; Pettit, G. R. Res. Inst. Integr. Sci., Kanagawa Univ. (Japan) 2000, 29-38. (e) Vu, H.; Ianosi-Irimie, M.; Danchuk, S.; Rabon, E.; Nogawa, T.; Kamano, Y.; Pettit, G. R.; Wiese, T.; Puschett, J. B. Exper. Biol. Med. 2006, 231, 215-220. (f) LaMarca, H. L.; Morris, C. A.; Pettit, G. R.; Nogawa, T.; Puschett, J. B. Placenta 2006, 27, 984-988. (g) Wang, J.-D.; Narui, T.; Takatsuki, S.; Hashimoto, T.; Kobayashi, F.; Ekimoto, H.; Abuki, H.; Ninjima, K.; Okuyama, T. Chem. Pharm. Bull. 1991, 39, 2135-2137. (h) Yamagishi, T.; Yan, X.-Z.; Wu, R.-Y.; McPhail, D. R.; McPhail, A. T.; Lee, K.-H. Chem. Pharm. Bull. 1988, 36, 1615-1617. (i) Kupchan, S. M.; Ognyanov, I.; Moniot, J. L. Bioorg. Chem. 1971, 1, 13-31.
- (9) (a) Ohta, G.; Koshi, K.; Obata, K. Chem. Pharm. Bull. 1968, 16, 1487–1497. (b) Smith, H. E.; Hicks, A. A. J. Org. Chem. 1971, 36, 3659–3668. (c) Smith, S. C.; Heathcock, C. H. J. Org. Chem. 1992, 57, 6379–6380. (d) Heathcock, C. H.; Smith, S. C. J. Org. Chem. 1994, 59, 6828–6839. (e) Pan, Y.; Merriman, R. L.; Tanzer, L. R.; Fuchs, P. L. Bioorg. Med. Chem. Lett. 1992, 2, 967–972. (f) Lietard, J.; Meyer, A.; Vasseur, J.-J.; Morvan, F. Tetrahedron Lett. 2007, 48, 8795–9798.
- (10) Liu, Z.; Meinwald, J. J. Org. Chem. 1996, 61, 6693–6699.
- (11) Cho, C.-G.; Park, J.-S.; Jung, I.-H.; Lee, H. Tetrahedron Lett. **2001**, 42, 1065–1067.
- (12) Scott, W. J.; Crisp, G. T.; Stille, J. K. J. Am. Chem. Soc. 1984, 106, 4630–4632.
- (13) (a) Misico, R. I.; Gil, R. R.; Oberti, J. C.; Veleiro, A. S.; Burton, G. J. Nat. Prod. **2000**, 63, 1329–1332. (b) Gupta, S.; Krasnoff, S. B.; Renwick, J. A. A.; Roberts, D. W.; Steiner, J. R.; Clardy, J. J. Org. Chem. **1993**, 58, 1062–1067. (c) Suenaga, K.; Kigoshi, H.; Yamada, K.

Tetrahedron Lett. 1996, 37, 5151–5154. (d) Sharma, P.; Powell, K. J.; Burnley, J.; Awaad, A. S.; Moses, J. E. Synthesis 2011, 2865–2892. (e) Wilk, W.; Waldmann, H.; Kaiser, M. Bioorg. Med. Chem. 2009, 17, 2304–2309.

- (14) (a) Josephsohn, N. S.; Snapper, M. L.; Hoveyda, A. H. *J. Am. Chem. Soc.* **2003**, 125, 4018–4019. (b) Danishefsky, S. J.; DeNinno, M. P. *Angew. Chem., Int. Ed. Engl.* **1987**, 99, 15–23. (c) Kerwin, J. F., Jr.; Danishefsky, S. *Tetrahedron Lett.* **1982**, 23, 3739–3742.
- (15) Gardi, R.; Castelli, P. P.; Gandolfi, R.; Ercoli, A. *Gazz. Chim. Ital.* **1961**, *91*, 1250–1257.
- (16) Kutney, J. P.; Piotrowska, K.; Somerville, J.; Huang, S.-P.; Rettig, S. J. Can. J. Chem. 1989, 67, 580–589.
- (17) The spectra of the other compounds characterized in this report are no longer available.